

Modeling chemical reactions with the Amsterdam Modeling Suite and the ReaxFF engine Chemical Vapor Deposition & Combustion



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ReaxFF

Reactive many-body interatomic potential

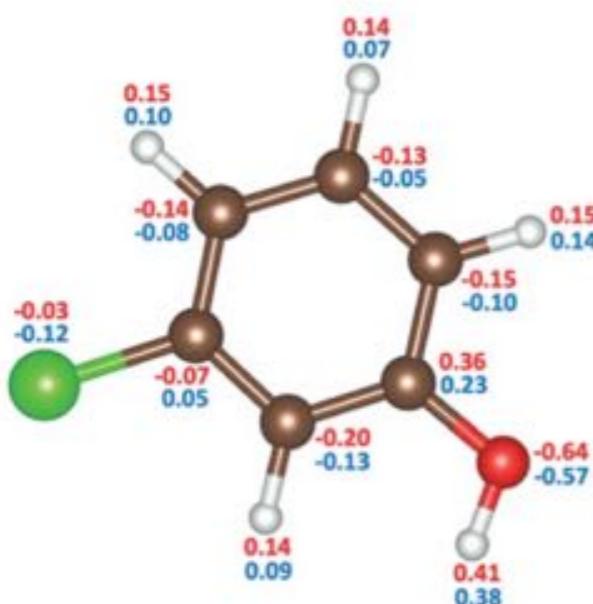
- The functional form

$$E_{pot} = \underbrace{E_{bond} + E_{ang} + E_{dih}} + E_{over/under} + E_{vdW} + E_{coul} + \dots$$

$$BO = \exp \left[p_1 \left(\frac{r}{r_0} \right)^{p_2} \right] + \exp \left[p_3 \left(\frac{r}{r_{0,\pi}} \right)^{p_4} \right] + \exp \left[p_5 \left(\frac{r}{r_{0,\pi\pi}} \right)^{p_6} \right]$$

Learn more: <https://www.scm.com/product/reaxff/>

- Partial atomic charges: QEq

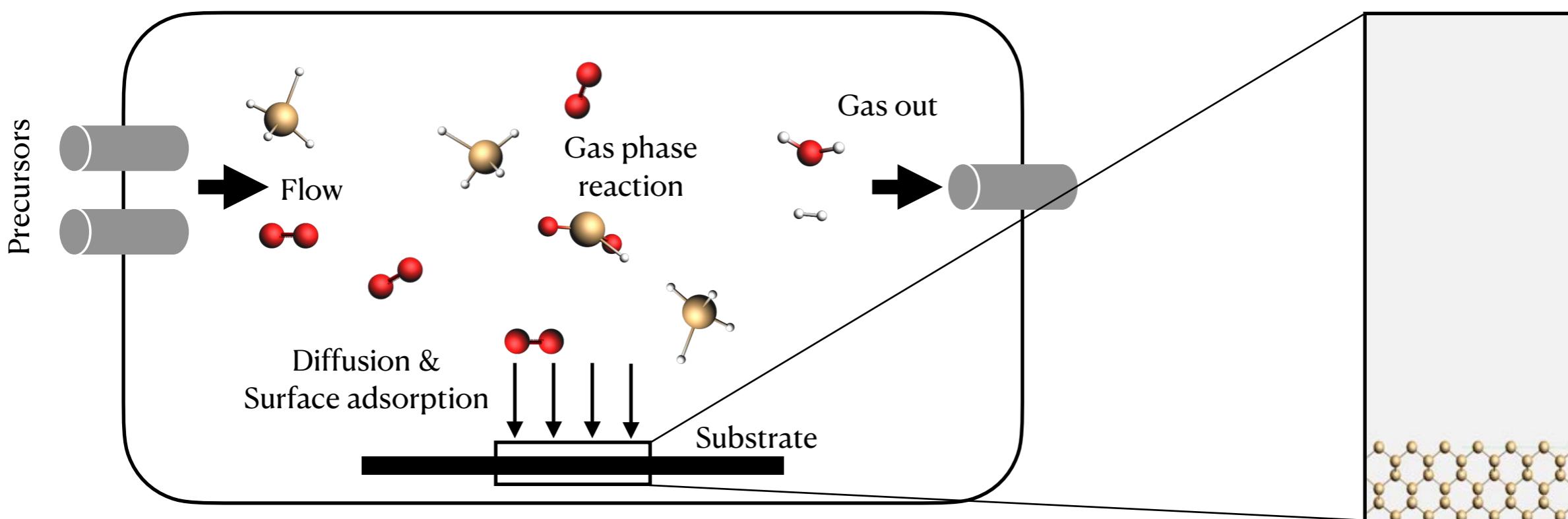


Chemical vapor deposition

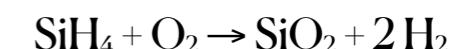
Principle of CVD

- To produce thin films in semiconductor industry
- Wafer exposed to gas/plasma which react/decompose on the substrate

<https://www.scm.com/doc/Tutorials/MolecularDynamicsAndMonteCarlo/MoleculeGunSimulationCVD.html>



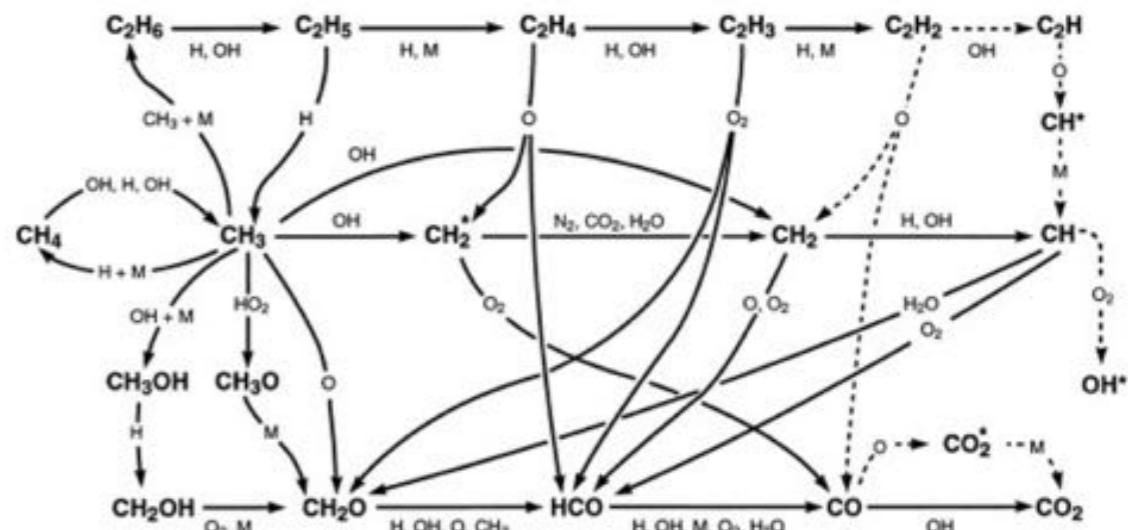
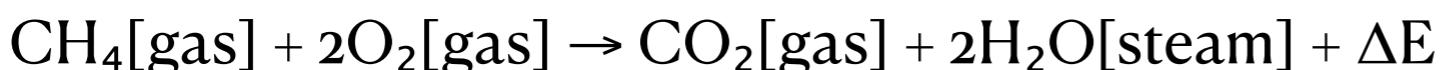
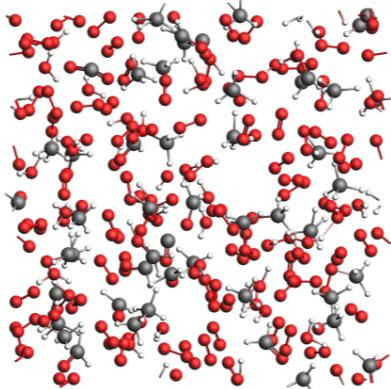
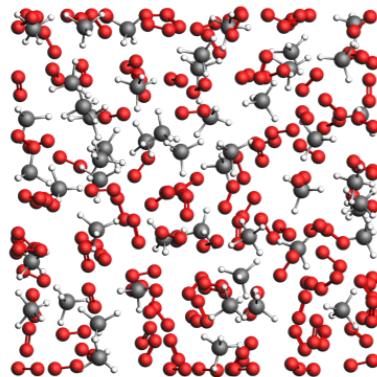
SiO₂ growth on Si substrate



Combustion reaction

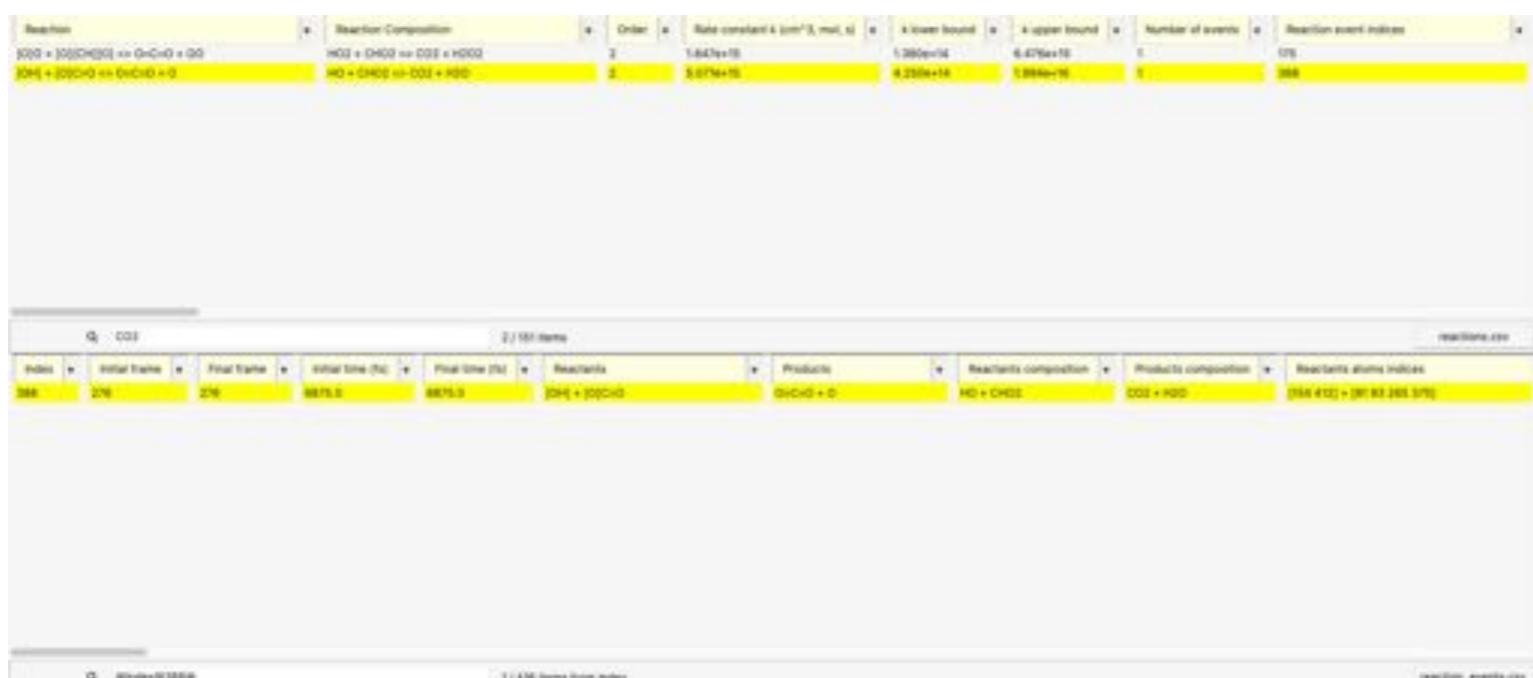
Combustion of methane in oxygen

- High T MD simulation of O₂ + CH₄



H.N. Najm, P.H. Paul, C.J. Mueller and P.S. Wyckoff,
Comb. Flame, 113, 312-332, 1998

- ChemTraYzer2



Reaction	Reaction Composition	Order	Rate constant k (cm ³ /s, mol, s)	k lower bound	k upper bound	Number of events	Reaction event indices
HO2 + [O]OCH2O2 <=> OH+O2 + O2O	HO2 + CH3O <=> CO2 + H2O	2	1.647e+10	1.380e+10	6.475e+10	1	175
HO + [O]OCH2O <=> HO2 + O2O	HO + CH3O <=> CO2 + H2O	2	8.074e+10	6.239e+10	1.289e+10	1	394

